

Smith and Wolff in 1969 reported the presence of a new hydroxy acid in seeds of Common thyme (Thymus vulgaris L.) (1). Chemical degradation studies and NMR spectroscopy demonstrated the identity of the compound as 2hydroxy-9(Z),12(Z),15(Z)octadecatrienoic acid (2hydroxylinolenic acid), and optical rotation data indicated that the chiral carbon had the "R" configuration. The mode of formation of 2-hydroxylinolenic acid in thyme seeds was not clarified, however, the presence of low levels of the odd-chain fatty acid 8(Z), 11(Z), 14(Z)heptadecatrienoic acid (norlinolenic acid) suggested that the plant α -oxidation pathway was involved.

The α -oxidation pathway in plants was characterized by Stumpf, who showed that preparations of peanut cotelydons catalyzed the oxidation of palmitic acid into a long-chain fatty aldehyde with concomitant liberation of CO₂ (2). In subsequent work, α oxidation of various C_n fatty acids into C_{n-1} aldehydes together with varying amounts of $C_n 2$ hydroxy acids and C_{n-1} fatty acids has been studied in preparations from higher plants and algae. The α -oxidation enzymes appear to operate together with aldehyde dehydrogenases and in this way provide a pathway for stepwise degradation of fatty acids into lower homologs. The involvement of 2-hydroperoxy fatty acids as intermediates in αoxidation was suggested by experiments by Shine and Stumpf, in which inclusion of a hydroperoxide reductant (glutathione/glutathione peroxidase) to an α -oxidation system led to decreased formation of aldehydes and CO₂ but increased formation of 2hydroxy acids (3). Conclusive evidence for 2-hydroperoxy fatty acids serving as intermediates in α -oxidation was provided ten

years ago by the actual isolation of 2-hydroperoxy fatty acids by two research groups. Thus, Akakabe and coworkers obtained 2(R)-hydroperoxypalmitic acid in incubations of palmitic acid with the α -oxidation system of the green alga Ulva pertusa (4), and Hamberg et al. obtained 2(R)-hydroxylinolenic acid in incubations of linolenic acid with α -oxidation enzymes in cucumber and with a new plantderived recombinant oxygenase (5). This enzyme was a hemecontaining fatty acid dioxygenase and was given the name "a-dioxygenase". Interestingly, α -dioxygenase from germinating pea was isolated as a dual function enzyme having a 70-kD subunit of α -dioxygenase-peroxidase and a 50-kD subunit of an NAD⁺dependent oxidoreductase (6).

On the basis of these findings the complete sequence of α -oxidation in plants can be formulated as:

 C_n fatty acid $\rightarrow C_n 2(R)$ hydroperoxy-fatty acid $\rightarrow C_{n-1}$ fatty aldehyde $\rightarrow C_{n-1}$ fatty acid

The first step is catalyzed by α dioxygenase whereas the second step can take place spontaneously because of the inherent chemical instability of the 2-hydroperoxides (*cf.* ref. 7). The third step is catalyzed by aldehyde dehydrogenases. It is thus clear that 2-hydroxy fatty acids are not intermediates in α oxidation but can be formed from the 2-hydroperoxy acids in a side reaction promoted by reductase or peroxidase enzymes.

Interestingly, α -dioxygenases are not only involved in the plant α oxidation pathway but also appear to provide products involved in plant physiology and plant pathology. Thus, the α oxidation pathway was activated when tobacco plants are infected with microbial pathogens, and 2hydroxy fatty acids exerted a tissue-protective effect in bacterially infected leaves (8). Additionally, a role of α dioxygenase in plants during drought and salt-stress has been proposed (9).

2(*R*)-Hydroxy-9(*Z*),12(*Z*),15(*Z*)octadecatrienoic acid (O-1803-17a) is isolated by Lipidox from a natural source. Other α oxidation products such as 8,11,14-heptadecatrienal and norlinolenic acid are also available. 1. Smith, C.R. and Wolff, I.A. (1969) Lipids 4, 9-14. 2. Stumpf, P.K. (1956) J. Biol. Chem. 223, 643-649. 3. Shine, W.E. and Stumpf, P.K. (1974) Arch. Biochem. Biophys. 162, 147-157. 4. Akakabe, Y. et al. (1999) Tet. Lett. 40, 1137-1140. 5. Hamberg, M. et al. (1999) J. Biol. Chem. 274, 24503-24513. 6. Saffert, A. et al. (2000) Plant Physiol. 123, 1545-1551. 7. Adam, W. et al. (1977) J. Amer. Chem. Soc. 99, 5768-5773. 8. Hamberg, M. et al. (2003) J. Biol. Chem. 278, 51796-51805. 9. Tirajoh, A. et al. (2005) J. Exp. Bot. 56, 713-723.